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On the unit cell of YCrO₃. By LEWIS KATZ, *Department of Chemistry, University of Connecticut, Storrs, Connecticut, U.S.A.*

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A finely powdered sample of YCrO₃ was recently prepared in these laboratories by Mr John T. Looby. Measurement of the X-ray powder pattern (Cu K α radiation) of this new compound revealed several relationships of the form $k_1q_1 = k_2q_2$, where the k 's are integers and the q 's are $\sin^2 \theta$ values (Hesse, 1948). Examples are:

$$0.0831 \equiv 2 \times 0.0417;$$

$$0.1283 \equiv 4 \times 0.0320;$$

$$0.2081 \equiv 5 \times 0.0417;$$

$$0.1751 \equiv 8 \times 0.0219.$$

The values for k_2 —2, 4, 5, 8—suggested that the crystals were tetragonal, but several attempts to index the photograph on the basis of a tetragonal unit cell failed. This was taken to mean that the squares of the lengths of at least two of the cell edges were sufficiently close to being commensurate, and the cell angles sufficiently close to 90°, to simulate the symmetry of the tetragonal system.

The Goldschmidt tolerance factor for yttrium chromium oxide is about 0.86 and would indicate an orthorhombic (or possibly monoclinic) distortion of the cubic idealized perovskite structure (Wells, 1950). Indeed, when the photograph was viewed uncritically, the lines seemed to form clusters in such a way as to resemble the photograph of a simple cubic compound. If the second strong cluster corresponded to the cubic 110 reflection, the cell would

Table 1 (cont.)

<i>I</i>	$\sin^2 \theta$		<i>hkl</i>
	Obs.*	Calc.	
<i>m</i>	0.1751	{ 0.1750 0.1757	410 303
<i>vw</i>	0.1804	{ 0.1807 0.1810	232 411
<i>vw</i>	0.1970	0.1972	402
<i>vw</i>	0.2005	0.2000	331
<i>s</i>	0.2081	{ 0.2083 0.2076	240 412
<i>w</i>	0.2130	{ 0.2124 0.2140	421 402
<i>w</i> ⁺	0.2247	0.2244	412
<i>vwv</i>	0.2301	—	—
<i>vw</i>	0.2341	0.2340	332
<i>m</i> ⁺	0.2389	0.2390	422
<i>m</i> ⁺	0.2458	0.2452	242
<i>s</i>	0.2555	{ 0.2558 0.2548	422 413
<i>m</i> ⁺	0.2913	0.2912	432
<i>vw</i>	0.3014	0.3024	250
<i>vw</i>	0.3122	{ 0.3122 0.3114	404 423
<i>w</i> ⁺	0.3227	0.3226	414
<i>m</i> ⁺	0.3317	0.3317	440
<i>vwv</i>	0.3383	{ 0.3384 0.3378 0.3392	433 441 252
<i>w</i>	0.3462	{ 0.3462 0.3458	441 404
<i>w</i>	0.3552	{ 0.3562 0.3562	414 531
<i>vw</i>	0.3669	{ 0.3672 0.3666	351 531
<i>w</i>	0.3811	{ 0.3812 0.3817 0.3805	442 532 610
<i>w</i>	0.4088	0.4090	612
<i>w</i> ⁺	0.4175	0.4173	260
<i>w</i>	0.4247	{ 0.4242 0.4255 0.4238	540 261 602
<i>vw</i>	0.4336	0.4342	612
<i>vw</i>	0.4420	{ 0.4423 0.4425	524 504
<i>w</i>	0.4594	{ 0.4594 0.4584	533 425
<i>w</i>	0.4642	0.4642	630
<i>vw</i>	0.4762	{ 0.4759 0.4758	542 361
<i>w</i>	0.4927	{ 0.4752 0.4926 0.4920	452 632 613
<i>w</i>	0.5124	{ 0.5120 0.5130	070 444
<i>w</i>	0.5194	0.5198	614
<i>w</i>	0.5733	{ 0.5734 0.5734 0.5743	462 454 703
<i>m</i> ⁻	0.5917	0.5910	642
<i>m</i> ⁻	0.6020	{ 0.6016 0.6014	624 722

* 114.6 mm. diameter Philips camera.

Table 1. *Monoclinic indexing of YCrO₃*

<i>I</i>	$\sin^2 \theta$		<i>hkl</i>
	Obs.*	Calc.	
<i>vwv</i>	0.0190	0.0195	$\bar{1}01$
<i>vwv</i>	0.0219	0.0217	101
<i>w</i>	0.0320	0.0322	111
<i>m</i>	0.0417	{ 0.0418 0.0411	020 200
<i>vw</i>	0.0452	—	—
<i>s</i>	0.0518	{ 0.0516 0.0521	210 120
<i>vw</i>	0.0637	{ 0.0640 0.0635	211 121
<i>vw</i>	0.0673	0.0677	220 β
<i>s</i>	0.0782	0.0780	202
<i>vvs</i>	0.0831	0.0829	220
<i>s</i>	0.0865	0.0866	202
<i>m</i>	0.0888	0.0885	$\bar{2}12$
<i>w</i>	0.0972	0.0971	212
<i>vw</i>	0.1063	0.1060	301
<i>vw</i>	0.1111	0.1100	$\bar{3}11$
<i>m</i>	0.1161	{ 0.1164 0.1157	311 131
<i>m</i> ⁻	0.1203	0.1198	$\bar{2}22$
<i>m</i>	0.1283	0.1284	222
<i>m</i>	0.1354	0.1351	230
<i>w</i> ⁺	0.1415	0.1414	321
<i>w</i>	0.1478	{ 0.1478 0.1475	321 231
<i>s</i>	0.1643	0.1645	400
<i>m</i> ⁺	0.1676	0.1672	040
<i>m</i> ⁻	0.1722	0.1720	232

have an edge of about 3.8 Å. The number of lines actually observed up to $\sin \theta = 0.8$ was roughly five times the number a 4 Å cubic cell could have produced in this interval. At higher values of $\sin \theta$ the lines were too diffuse to measure or even count.

The following observations were sufficient to select a unit cell. The 'cubic 200' reflection was split into two lines with the one of smaller Bragg angle more intense than the other; this was accounted for by assuming the a and c axes about equal and the b axis somewhat shorter. The 'cubic 110' reflection appeared split into four lines consisting of one line and a close doublet on opposite sides of a very strong line; these were accounted for by also taking β different from 90° . Weak lines corresponding to spacings greater than 4 Å were present; these were accounted for by doubling the cell edges. The resulting cell had the following dimensions:

$$a \approx c = 7.61 \pm 0.01, \quad b = 7.54 \pm 0.01 \text{ \AA}; \quad \beta = 92^\circ 56' \pm 6'.$$

All but two very weak lines were indexable with this cell. Indices were assigned on the basis of Bragg-angle agreement with no consideration as to the possible space group. Except for 504 and 070, however, all the indices are consistent with $P2_1/n$. Both of these exceptions are ambiguous, and, in view of the discussion by Vousden (1951), should be considered less probable than their listed alternates (Table 1).

Acta Cryst. (1955). 8, 122

A rapid numerical method of calculating structure factors. By E. STANLEY,* *Division of Physics, National Research Council, Ottawa 2, Canada*

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Simple, rapid methods of calculating structure factors usually depend on the use of some geometrical, mechanical or electrical analogue. The methods of Vand (1948), Llewellyn (1951), Phillips (1952), Stanley (1952) and Grenville-Wells (1954) all suffer from some disadvantages such as limited accuracy, the necessity of preparing preliminary tables of $(hx+ky)$, or the inconvenience of the presentation of the result as the length of a line or the deflection of a galvanometer. Numerical methods for calculating the product $\frac{\cos(hx)}{\sin(hx)} \frac{\cos(ky)}{\sin(ky)}$ are convenient only when a single product is required. The method here described is more accurate than most analogue devices and is numerical.

The basis of the method is a single table of values of $\cos x$ at intervals of 0.001 in x , arranged in a rectangular array so that successive rows overlap. In the present table each value of $\cos x$ is contained in a rectangle 1.0 cm. \times 0.5 cm. and the values of x in successive rows run thus:

$$0.000-0.040, 0.020-0.060, 0.040-0.080, \dots, 0.980-0.020, \\ 0.000-0.040, 0.020-0.060, \dots, 0.980-0.020.$$

Positive and negative values of $\cos x$ are distinguished by the use of different coloured inks and the values of

If a is not exactly equal to c (they are not likely to differ by more than 0.015 Å however), the crystals are probably monoclinic. If a is exactly equal to c , choosing the diagonals of the a - c face as new a and c axes results in a cell which is orthogonal, B centered, and has cell edges of 10.49, 7.54 and 11.04 Å. This cell would imply that the crystals are orthorhombic. The orthorhombic a and c axes could be halved if the monoclinic cell were end centered, but the 332 and $\bar{5}24$ or 504 reflections, both very weak, prevent this possibility.

Another possibility is that the cell is similar to that of NaNbO_3 (Vousden, 1951). In this case we would expect the dimensions of the orthorhombic cell to be 5.24, 15.08 and 5.52 Å. If one takes into account the ambiguities in some of the indexing, no reflections prevent this possibility. In fact, this cell could also account for the two previously non-indexed reflections. A decision as to which unit cell is correct, as well as any detailed evaluation of the structure, will probably have to await the growing of satisfactory single crystals.

References

- HESSE, R. (1948). *Acta Cryst.* 1, 200.
 VOUSDEN, P. (1951). *Acta Cryst.* 4, 545.
 WELLS, A. F. (1950). *Structural Inorganic Chemistry*. Oxford: Clarendon Press.

x for both $\sin x$ and $\cos x$ are marked along the edges of the table. This type of arrangement of four complete cycles is necessary in order that the masks, described below, can be moved over the table and the correct relationship between the positions of the values of $\cos x$ preserved. The particular arrangement is flexible and the present table, which can be mounted on a board 18 in. \times 20 in., has been designed for convenient desk use.

Masks are prepared, which cover the area of one cycle of the table, from which are cut rectangular holes 1.0 cm. \times 0.5 cm. at positions corresponding to ky . One mask is required for each value of y punched with holes corresponding to successive orders of k up to, say, 20, and the holes are marked with the values of k . On the scale proposed the masks are approximately quarto size.

To use the method a table of values of hx is prepared for each atom and the appropriate mask for the y value of the first atom is selected. The mask is placed with the zero-order hole at 000 in the table. The values of $\cos ky$ appear in the holes marked k from 0 to 20. These values are conveniently noted on a recording adding machine. The mask is then moved until the zero-order hole is at x in the table when all the values of $\cos(x+ky)$ appear in the holes marked k and are similarly recorded. This procedure is repeated with the zero-order hole at $2x$, $3x$, etc. until all the values of $\cos(hx+ky)$ have been recorded. Negative values of k can be obtained by punching

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